

## Conversion of CO<sub>2</sub> using electrochemical flow processes

Damien VOIRY - *University of Montpellier, France*

The conversion of CO<sub>2</sub> *via* electrochemical processes is a relevant technology to close the carbon cycle; especially when combined with renewable energy sources. Because of their high market value and their high energy density, research has aimed at developing catalysts for the electrochemical conversion of CO<sub>2</sub> into multicarbon molecules. Copper (Cu) is one of the few transition metals that can efficiently catalyze the electrolysis of CO<sub>2</sub> to multicarbon products such as ethylene, ethanol, acetate, propanol. The design of Cu-based catalysts by adapting some of the concept of molecular catalysts in order to finely tailor the behavior of the active sites of metallic surfaces is currently regarded as the long-standing interest for the controlled design of novel electrocatalytic materials. Increasing the oxidation state of copper has been suggested to improve the CO<sub>2</sub>RR performance and notably the formation of C<sub>2+</sub> species.

In this context, we have proposed a new strategy to improve the conversion of CO<sub>2</sub> to hydrocarbon molecules with two or more carbon atoms (C<sub>2+</sub>) via molecular doping of a metal catalyst. Specifically, we have identified electrophilic functional groups that can direct the electrochemical reactions towards the production of C<sub>2+</sub> species such as ethanol and ethylene and improve the reaction rates at the catalyst surface. Our research has also focused on the integration of our electrocatalyst into a flow electrolysis process. Flow electrolyzers have recently been proposed to facilitate electrochemical CO<sub>2</sub>RR due to their unique ability to achieve electroreduction at high reaction rates via the creation of a three-phase interface. Although some examples of flow electrolyzers for CO<sub>2</sub> conversion have been recently reported, the influence of CO<sub>2</sub> and electrolyte flows on the overall catalytic mechanism has remained ambiguous. We explored the correlation between the applied potential and the feeds in both electrolyte and CO<sub>2</sub>, on the one hand, and the performance metrics, on the other hand.

I will review our recent findings on our recent progress in electrocatalytic CO<sub>2</sub> reduction using a continuous flow process.